Cytotoxic Long-chain Alkene and Terpene Isolated from the Methanol Extract of the Air-dried Leaves of

Pipturus arborescens C.B. Rob

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Abstract

A long-chain alkene (1-hexacosene) and a terpene were obtained from the bioactive methanol extract of the air-dried leaves of *Pipturus arborescens* (Link) C.B. Rob ("Handalamay") collected from Poblacion, Kapatagan, Lanao del Norte. The compounds were isolated through extraction with ethanol, sequential partitioning with water-chloroform and hexane-90% methanol followed by fractionation and purification with repeated gravity column chromatography using gradient mixtures of hexane: ethyl acetate (1:1 ratio). The structures of the isolates were elucidated on the basis of spectroscopic data: ¹H-and ¹³C-NMR, ¹H-¹³C Heteronuclear Multiquantum Correlation (HMQC), ¹H-¹³C Heteronuclear Multipode Correlation (HMBC), ¹H-¹H Homonuclear Correlation Spectroscopy (UV-VIS) and Mass Spectrometry (MS). The isolated compounds were shown to possess moderate cytotoxic activity against the brine shrimp *Artemia salina* (Leach).

Keywords: bioactive, chromatography, fractionation, spectroscopy,1-hexacosene

Introduction

Plants have been the basis of many traditional medicine systems throughout the world for thousands of years and continue to provide mankind with new remedies. Organic compounds from terrestrial plants have extensive past and present uses in treatment for many diseases and serve as compounds of interest both in their natural form as well as templates for synthetic modification. The practical use of plants for health care provides a substantial resource in meeting drug requirements. However, the prevalent medicinal use of plants is mostly based on tradition. Thus, the need for scientific validation is invoked.

Pipturus arborescens (Link) C.B. Rob ("Handalamay" in Visayan and "Dalunot" in Tagalog) is a widely distributed species in the Philippines that belongs to family Urticaceae. A recent study reported that methanol extract of certain species belonging to Urticaceae family exhibited cholinesterase inhibitory activity (Dhivya et al., 2014). Another study also showed that the hydroalcoholic extract of a plant under Urticaceae family has hypoglycemic effect and protective activity of beta-cells of Langerhans in hyperglycemic rats (Golalipour et al., 2007). P. arborescens is used as a remedy for boils and skin diseases known as Herpes zoster ("ugahip" in Visayan) (Quisumbing, 1978). The scrapings of the bark are used externally as a cataplasm for boils while the leaves are used to cure herpes (Esperanza & Kitche, 2005). A study reported that triterpenes such as glutinone, friedelin, glutinol, and a mixture of common sterols such as campesterol, stigmasterol and sitosterol have been isolated from the hexane extract of the leaves (Gabona, 2000). Another study also reported that ficaprenol-10 and squalene were also isolated from methanol extract of *P. arborescens* the leaves (Peteros, 2010).

It has also been reported that a pure isolate obtained from the crude ethyl acetate extract of the leaves of *P. arborescens* is active against *Bacillus subtilis* (Rosal, 1995). A preliminary study revealed that the methanolic extract of the air dried leaves of *P. arborescens* exhibited significant activities against *Staphylococcus aureus* and *Escherichia coli* (Enerio, 2007). Thus, the present study was conducted in order to further investigate the possibility on the isolation, purification and identification of the chemical components from the bioactive fraction of air-dried leaves of *P. arborescens* using chromatographic methods, and to elucidate the molecular structures through various spectroscopic methods, and finally to determine the cytotoxic activity of the isolates using the Brine Shrimp Lethality assay.

Materials and Methods

General

Column chromatography was carried out using silica gel 60 (Merck, 70-230 mesh, ASTM) as an adsorbent. All NMR spectra were recorded on a JEOL AL- 400 (400 MHz⁻¹H NMR, 100 MHz⁻¹³C NMR) spectrometer in CDCl₃ solutions using tetramethylsilane (TMS) as an internal standard. MS data were recorded on a Shimadzu LC-MS MALDI-TOF[MS]. UV spectrometric profiles of the isolates were obtained using the Shimadzu 160A UV-Vis Spectrophotometer at the wavelength range of 200-800 nm with methanol as the reference solvent. The IR spectra were recorded on a Horiba FT-IR 720 spectrometer.

Sampling and Sample Preparation

Mature leaf samples of *P. arborescens* were collected and obtained from Kapatagan, Lanao del Norte, Philippines. The samples were air-dried for one month, cut into small pieces, ground in a grinder to obtain a fine powder and stored in polyethylene bags at room temperature. Plants were authenticated by Professor Josefa D. Villanueva of the Department of Biological Sciences, College of Science and Mathematics, MSU-Iligan Institute of Technology.

Phytochemical screening

The ground sample (1.06 kg) of air-dried leaves was soaked with 95% ethanol for seven days in the refrigerator at 5°C. The ethanol extract was then filtered and concentrated under a rotary evaporator at a temperature below 40° C to yield 30.99 g of crude ethanolic extract. The crude ethanol extract was then partitioned between water and chloroform to give the aqueous extract, the chloroform extract and a considerable amount of brick-red solid interphase. A 14.50 g portion of chloroform extract was subsequently partitioned between hexane and 90% methanol to give 5.15 g of hexane extract and 4.91 g of methanol extract. Methanol was chosen for further study since it has been reported to have the most

significant antimicrobial activity among the four extracts (Enerio, 2007). Methanol extract was eluted through a silica gel column using gradient mixtures of hexane-ethyl acetate and methanol-ethyl acetate at 10% increment then 10% methanol in chloroform and 10% water in methanol. Thin Layer Chromatography (TLC) was used to characterize and pool the eluted fractions. A sub-fraction from the methanol extract was then further purified through a series of three column chromatography using gradient mixtures of hexane-ethyl acetate to obtain two isolates.

Brine Shrimp Toxicity Assay

Natural seawater was used as the culture medium. Seawater was boiled for 30 minutes and then filtered to obtain a sterile medium. The medium was poured into the hatching tank and about 100 to 200 mg cysts were introduced. A 100-watt lamp was positioned to provide direct light and warmth throughout the embryogenesis. After 36 hours, the nauplii were collected using Pasteur pipette from the hatching tank and were transferred to a Petri dish with 5.0 mL of seawater. The assay was carried out according to the principle and protocol previously described by Meyer et al. in 1982 with slight modifications. Ten nauplii were transferred to a sample test tube and sterilized seawater was then added to make a volume of 5.0 mL. A parallel series of tests with the positive control (podophyllotoxin) and a blank control (methanol) was also conducted. There were three replicates in each treatment. Number of survivors and percentage lethality were evaluated after 6 hours and 24 hours.

Results and Discussion

Two isolates, namely Isolate 1 and Isolate 2 were obtained from the methanol extract of the air-dried leaves of *P. arborescens*. Isolate 1 was obtained as colorless oily liquid yielding only 25.20 mg mass. The ¹H and ¹³C signals for Isolate 1 are summarized in Table 1. Its molecular formula was established to be $C_{26}H_{52}$ on the basis of the LC-MS data wherein the compound displayed an ion peak at 365.1 corresponding to $[M^++ H]$. The FT-IR spectrum obtained on thin film exhibited absorption bands at 3077.83, 2923.56, 2854.13 cm⁻¹ (sp²-CH and sp³-CH stretching) and 1639.20-1720.19 cm⁻¹ (C=C stretching). The UV spectrum showed the absorption maximum at 211 nm characteristic of an alkene system. The ¹³C NMR and ¹³C DEPT spectra showed the presence of an olefinic terminal methylene carbon (δ 114.0 ppm), an olefinic methine carbon (δ 139.2 ppm), a cluster of methylene carbons (δ 22.8-33.9 ppm) and a methyl carbon (δ 14.2 ppm). The ¹H-NMR spectrum displayed a signal at δ 5.81 ppm attributed to the olefinic methine proton. The doublet of doublets signals at δ 4.99 ppm (J_{11} = 17.2, 1.4 Hz) and at δ 4.92 ppm (J_{11} = 10.3, 1.4 Hz), are assigned to the terminal olefinic methylene protons. The signal at δ 2.04 ppm is due to the methylene protons next to the olefinic methine proton. The broad signals at δ 1.26 – δ 1.37 ppm are typical of internal methylene protons while the triplet (J=3.0 Hz) signal at δ 0.88 corresponds to the lone methyl protons.

Position No.	δ ¹ H ^a , ppm	Multiplicity (J in Hz)	δ ¹³ C ^b , ppm
1	4.99 4.92	dd (17.2, 1.4) dd (10.3,1.4)	114
2	5.81	m	139.2
3	2.04	m	33.9
4	1.26-1.37	broad m	22.8-32.0
5	0.88	t (3.0)	14.2

 Table 1. ¹H-and ¹³C-NMR Data of Isolate 1.

^aData measured in CDCl₃ at 400 MHz.

^bData measured in CDCl₃ at 100 MHz.

In the COSY spectrum, the terminal olefinic methylene protons (δ 4.99, 4.92) were coupled to the olefinic methine proton at δ 5.81 which in turn showed coupling with the methylene signal at δ 2.04. The methylene protons at δ 2.04 then coupled with the broad methylene peaks at δ 1.26-1.37 which showed coupling with the methyl protons (δ 0.88). With the corroboration of the LC-MS, IR, UV, and NMR data, the structure of Isolate 1 could be proposed as 1-hexacosene (Figure 1).

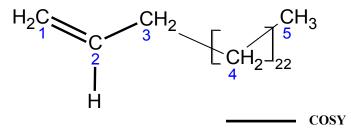


Figure 1. Proposed structure of Isolate 1 (1-hexacosene).

Isolate 2 was obtained as light orange oil which gave a lighter pink color under UV lamp and brown color upon exposure to iodine vapor. It yielded 40.20 mg of the product. The ¹H-and ¹³C-NMR Data of Isolate 2 are summarized in Table 2. The FT-IR spectrum of Isolate 2 obtained on thin film showed absorption bands at 2726.85-2958.27 cm⁻¹ (sp² and sp³ C-H stretch), 1373.07-1454.06 (C-H bending of a methyl group), and 1592.91-1668.12 (C=C stretching). The UV spectrum displayed absorption maxima at 212, 254 and 284 nm indicating the presence of a polyene system. The ¹³C-NMR spectrum indicated the presence of quaternary carbons (δ 131.1, 135.2 ppm), olefinic methine carbons (δ 124.1, 124.3, 124.4, 124.8, 124.9 ppm), methyl carbons (8 16.2, 17.8, 22.8, 23.3, 23.4, 25.7, 26.6 ppm), methylene carbons (\$ 24.9, 25.8, 26.4, 26.5, 26.8, 32.1, 32.3, 39.8 ppm) and terminal methine carbon (δ 17.8 ppm). In the COSY spectrum, the olefinic methine proton at δ 5.11 ppm was coupled to the methylene protons at δ 2.0-2.1 ppm which in turn showed coupling to the methyl protons at δ 1.60, 1.68, 1.74 ppm. The methylene protons also showed another coupling to the terminal methine proton at δ 1.56 ppm which displayed coupling to the methyl protons at δ 1.28 and 0.86 ppm.

Position Number	δ ¹ H ^a , ppm	δ ¹³ C ^b , ppm
1	1.56	17.8
2	2.04-2.16	32.1-32.3
3, 7	5.11	124.1-124.4
4	-	131.1-135.2
5	2.04-2.16	39.8
6	2.04-2.16	26.8
8	-	131.11
9	1.68	25.7
10	1.28	26.6
11	0.86	22.8
12	1.6	16.2
13	16	17.8

^aData measured in CDCl₃ at 400 MHz

^bData measured in CDCl₃ at 100 MHz

The HMQC and the HMBC correlations, made possible the assignment of the chemical shifts of protons directly attached and correlated two to three bonds away, respectively to a given carbon atom. The proton signal at δ 1.68 ppm is correlated two bonds away to the quaternary carbons (δ 131.1, 135.2 ppm) and to the olefinic methane carbons (§ 124.1, 124.3, 124.4 ppm). This proton signal is also correlated three bonds away to the methylene carbons at δ 32.3 ppm. The methyl proton at δ 1.60 ppm displayed a correlation with the quaternary carbon at δ 135.2 ppm. Meanwhile, the methylene protons (δ 2.0-2.1 ppm) exhibited correlation to the olefinic methine carbons (δ 124.1, 124.3, 124.4 ppm), to the quaternary carbons two and three bonds away (δ 131.1, 135.2 ppm) respectively, and to the methylene carbons (δ 32.1, 32.3, 39.8 ppm). The terminal methyl protons (δ 1.28, 0.86 ppm) are correlated to the methylene carbon at δ 32.3 ppm. Furthermore, the methyl protons at δ 0.86 also showed a correlation to the terminal methyl carbon at δ 26.6 ppm. The important confirmation regarding the mass of the molecular ions and hence the exact molecular weight of Isolate 2 could not be established as there is no available mass spectral data. However, on the basis of IR, UV, and NMR data, the structure of Isolate 2 could be proposed as a terpene (Figure 2).

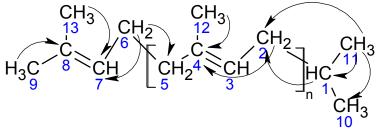


Figure 2. Proposed structure of Isolate 2 (Terpene).

The LC_{50} of Isolate 1 was not determined as the amount of the compound extracted was not sufficient enough to conduct the experiment. However, its cytotoxicity was evaluated by determining percent mortality rate at 1000 ppm concentration, and was found to be 40.00%. The mortality of brine shrimp assay in Isolate 2 is given in Table 3.

Concentration (ppm)	Mortality (%)
10.0	0.00
100.0	10.20
1000.0	45.71

Table 3. Mortality of the brine shrimp larvae after 24 h of exposure to Isolate 2.

The results indicate that Isolate 2 exhibits the mortality of 45.71% against the test animal at 1000.0 ppm, significantly low activity (10.2%) at 100.0 ppm and no activity at all at low concentrations such as 10.0 ppm. In the estimation of LC_{50} using Probit analysis (Finney, 1971), it could be suggested that. Isolate 2 has LC_{50} of 2344.23 ppm with a standard error of 1.74 at 95% confidence level, thus the range could be suggested as 2340.75-2347.71 ppm. The BSLT assay has proven to be a convenient system for screening bioactive natural products. It has the complete and effective range to test the toxicity as well as for various assays (Kanwar, 2007). The brine shrimp lethality assay represents a rapid, inexpensive and simple bioassay for testing plant extracts bioactivity which in most cases correlates reasonably well with cytotoxic and anti-tumor properties (McLauglin et al., 1993). The cytotoxicity of Isolate 2 provides baseline information that this is possible for clinical use indicative of anticancer, antiviral and other wide-range of pharmacological activities (Saveed et al., 2003). The exact mechanism of cytotoxic action of any tested isolates is not yet known; it is simply provided by disturbing the fundamental mechanisms concerned with cell growth, mitotic activity, differentiation and function (Goodman et al., 1980). However, to better evaluate the potential cytotoxic effectiveness of Isolate 2 from the leaves of P. arborescens, it warrants further investigation and specific studies.

Conclusion and Recommendation

The bioactive methanol extract of the air-dried leaves of *Pipturus arborescens* (Link) C.B. Rob ("Handalamay") afforded a long-chain alkene and a terpene through repeated gravity column chromatography. The isolated compounds exhibited moderate toxicity towards the brine shrimp and it can be inferred that the isolated compounds are biologically

active. With more advanced chromatographic techniques, there are certainly many compounds with interesting activities yet to be discovered from *P. arborescens*. Extracts, fractions and the isolated compounds should be subjected to as wide a range of bioassays as possible in order to confirm this conclusion.

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